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In chemistry, pH (/ p i? ? e? t? /, denoting 'potential of hydrogen' or 'power of hydrogen') is a scale used to specify the acidity or basicity of an aqueous solution. Acidic solutions (solutions with higher concentrations of H⁺ ions) are measured to have lower pH values than basic or alkaline solutions.. The pH scale is logarithmic and inversely indicates the concentration of hydrogen ...

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Acid strength is the tendency of an acid, symbolised by the chemical formula HA, to dissociate into a proton, H⁺, and an anion, A⁻. The dissociation of a strong acid in solution is effectively complete, except in its most concentrated solutions.. HA ? H⁺ + A⁻. Examples of strong acids are hydrochloric acid (HCl), perchloric acid (HClO₄), nitric acid (HNO₃) and sulfuric acid (H₂SO₄).

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Measurements at extreme pH values and temperatures, or low conductivity may take longer than those of aqueous solutions at room temperature with a neutral pH value. How long does a pH probe last? An inline pH probe lasts for varying amounts of time, depending on the conditions surrounding your in-line pH analyzer.

[ph indicator strip | Sigma-Aldrich](#)

Albumin, Bovine Serum, 10% Aqueous Solution, Nuclease-Free. 1 Product Result | Match Criteria: CAS Number, Related Cas Number